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# MULTIPLE FORMS OF 5'-NUCLEOTIDASE FROM LACTATING RAT MAMMARY GLAND RESULTING FROM THE ASSOCIATION OF THE ENZYME WITH DIFFERENT MEMBRANE FRACTIONS

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# **Summary**

Microsomes isolated from rat mammary gland can be separated into two fractions by flotation on a discontinuous sucrose density gradient. The lighter fraction (F<sub>1</sub>) exhibits by Lineweaver-Burk analysis two forms of 5'-nucleotidase (5'-ribonucleotide phosphohydrolase, EC 3.1.3.5), a low  $K_{\rm m}$  form  $(K_{\rm m}, 22 \,\mu{\rm M})$ and a high  $K_{\rm m}$  for  $(K_{\rm m}, 133 \, \mu {\rm M})$ . The heavier fraction  $(F_2)$  shows only the low  $K_{\rm m}$  form. Digitonin treatment of  $F_1$  followed by reflotation permits an additional fractionation (Huggins, J.W. and Carraway, K.L. (1976) J. Supramol. Struct. 5, 59-63). The digitonin-shifted heavier fraction  $(F_1DF_3)$  is enriched in plasma membrane markers, is depleted of galactosyltransferase and contains the high  $K_{\rm m}$  form of the nucleotidase. The unshifted fraction  $(F_1DF_3)$  is enriched in galactosyltransferase, depleted in plasma membrane markers and contains the low  $K_{\rm m}$  form of the nucleotidase. Digitonin treatment and reflotation of  $F_2$ permits isolation of a density-shifted fraction (F<sub>2</sub>DF<sub>3</sub>) which is enriched in plasma membrane markers and galactosyltransferase and which contains the unchanged low  $K_{\rm m}$  form of the nucleotidase. Solubilization of either  $F_1$  or F<sub>1</sub>DF<sub>3</sub> with deoxycholate or Triton X-100 results in conversion of essentially all of the nucleotidase to a low  $K_{\rm m}$  form.

Arrhenius plots of all forms of the enzyme (high  $K_{\rm m}$ , low  $K_{\rm m}$ , digitonin treated, solubilized) exhibit a temperature transition in the 15–20°C range. In addition membranes exhibiting high  $K_{\rm m}$  forms have a transition near 35°C which is abolished upon treatment with detergents. Inhibition of the low  $K_{\rm m}$  form of nucleotidase by concanavalin A is highly cooperative (Hill coefficient, 1.9–2.4). The high  $K_{\rm m}$  form shows less cooperativity (Hill coefficient, 1.4) in

untreated membranes, but a pronounced increase in cooperativity is observed upon solubilization in detergents.

We propose that the two enzyme forms are best explained by differences in the association of a single type of enzyme molecule with the different types of membranes. According to this proposal specific membrane enzyme-protein or enzyme-lipid interactions place restrictions on the nucleotidase which result in the higher  $K_{\rm m}$ , decreased cooperativity and new thermal transition observed for the enzyme of the  $F_1$  and  $F_1\mathrm{D}F_3$  fraction.

#### Introduction

Although the 5'-nucleotidase (5'-ribonucleotidase phosphohydrolase, EC 3.1.3.5) of mammalian cells has received a considerable amount of attention in recent years, its function is still unclear [1]. The enzyme is localized predominantly at the cell surface in a number of cell types [2—7]. Although it is usually treated as a single species in assessing membrane isolation and purification, there is a considerable amount of disparity in the properties of the enzyme from different sources [8]. Evidence for multiple forms of the enzyme has been presented in the case of the bovine milk fat globule membrane [9].

Recently we have demonstrated that 5'-nucleotidase can be used as a reporter group to monitor changes which occur at the cell surface which result from cellular perturbations [10]. For example, the treatment of intact 13762 mammary ascites tumor cells with colchicine, cytochalasins, dibucaine or the ionophore A23187 in the presence of  $Ca^{2+}$  causes an increase in the cooperativity observed for the inhibition of the ecto-nucleotidase by the plant lectin concanavalin A (Ref. 10 and unpublished data). The cooperativity change appears to be related to the mode of association of the enzyme with the membrane. Extraction of isolated ascites plasma membranes under conditions that remove substantial amounts of cytoskeletal proteins (e.g. actin,  $\alpha$ -actinin and actin binding protein) [11] causes the inhibition of the membrane nucleotidase by concanavalin A to become cooperative [7].

Such results indicate a need to understand the association of nucleotidase with the membrane and the influence of that association on the properties of the enzyme. Toward this end we have studied a different system, plasma membranes isolated from lactating rat mammary gland. In previous studies on isolation and characterization of plasma membranes from rat mammary gland we noted variations in the kinetics of the nucleotidase depending on the membrane fraction and assay techniques used (Carraway, C.A.C., unpublished data). We now report further analysis, which has shown the presence of two enzyme forms, the apparent  $K_{\rm m}$  values of which are separated about five-fold. The two forms segregate with different membrane fractions which can be separated by density perturbation techniques. The two forms also differ in the cooperativity of a temperature transition of the activity at about 35°C. All of these differences can be eliminated by solubilizing the membranes in deoxycholate or Triton X-100, suggesting that they are related to the mode of association of the enzyme with the membrane.

A preliminary account of a part of this work was presented previously [12].

# **Experimental procedures**

Materials. Partially purified rat mammary plasma membranes ( $F_1$  and  $F_2$ ) were obtained as previously described [13,14], and subjected to further fractionation by the digitonin shift procedure to yield a highly purified plasma membrane fraction and a Golgi fraction [14].  $F_1$  and  $F_1DF_1$  fractions were obtained from the 0.9%-32% sucrose interface,  $F_2$  from the 32%-36% sucrose interface and  $F_1DF_3$  and  $F_2DF_3$  from the 36%-40% interface (Ref. 14 and unpublished data). Concanavalin A, enzymes, substrates and deoxycholate were from Sigma Chemicals.

Analytical procedures. For the assay of 5'-nucleotidase, membranes or solubilized fractions at an appropriate concentration were suspended in assay buffer (50 mM Tris-HCl, 1 mM MgCl<sub>2</sub>, pH 8.0) and incubated for 10 min at 37°C. Subsequently  $\beta$ -glycerophosphate was added to 5 mM prior to initiation of the reaction. The reaction was started by adding AMP (containing 1–10  $\mu$ Ci/ml of [<sup>32</sup>P]AMP) to make a total assay volume of 0.4 ml. The reaction was run at the appropriate temperature and terminated by adding 1 ml of ice-cold 10% (w/v) trichloroacetic acid solution containing 10% (w/v) acid-washed charcoal. The assay tubes were then centrifuged at 5000 × g for 15 min to sediment the charcoal, and a 500  $\mu$ l aliquot of the supernatant in Aquasol-2 was counted in a Packard Tri-Carb Model 3320 Liquid Scintillation Counter. Blanks were made in which no enzymes were added.

Protein was determined by the method of Lowry et al. [15].

Concanavalin A inhibition experiments were performed as previously described [7,16].

### Results

Multiple enzyme forms in purified mammary membrane fractions

Plasma membrane-enriched material is obtained in two fractions ( $F_1$  and  $F_2$ ) after flotation of a microsomal preparation on a discontinous sucrose density gradient [13]. When the partially purified light rat mammary plasma membrane fraction ( $F_1$ ) was used, complex kinetic behavior was observed for the nucleotidase. Lineweaver-Burk plots showed two distinct straight lines (Fig. 1), suggesting the presence of two forms of the enzyme. When the heavy plasma membrane fraction ( $F_2$ ) was examined, an essentially hyperbolic response was observed for the substrate concentration dependence of the nucleotidase velocity. The Lineweaver-Burk analysis (Fig. 2) gave a  $K_m$  of 27  $\mu$ M, essentially identical to the low  $K_m$  form of the enzyme in  $F_1$ . However, it should be noted that the Lineweaver-Burk plot shows a slight 'tailing' at the high substrate end, suggesting the presence of a small amount of enzyme with a higher  $K_m$ . This is not too surprising, since the  $F_2$  fraction, upon recentrifugation, will reequilibrate to yield some  $F_1$  (unpublished data).

The results above suggest that there are two forms of the nucleotidase present in the partially purified  $F_1$  fraction. Since this fraction contains two different types of membranes, which can be separated by digitonin treatment and flotation on a density gradient [19], we examined the separated subfractions  $(F_1DF_1$  and  $D_1DF_3)$  to determine whether the two enzymes forms are

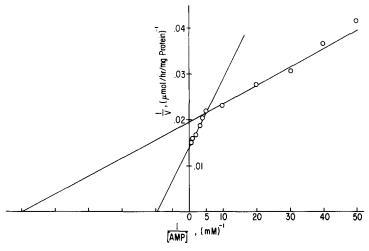


Fig. 1. Lineweaver-Burk plot of  $F_1$  membrane fraction 5'-nucleotidase kinetics. Membranes were prepared, incubated and assayed as in Experimental procedures. Membrane protein concentration was 22  $\mu g/assay$ .

segregated in these two types of membranes. The digitonin treatment of the F<sub>1</sub> fraction did not significantly alter the kinetic behavior of the enzyme (Table I).

When the purified fractions  $F_1DF_1$  and  $F_1DF_3$  are characterized separately, they show clearly different kinetic plots, indicating that each form of the enzyme is present in a different type of membrane. The low  $K_m$  form is associated with the putative Golgi fragments  $(F_1DF_1)$ , while the high  $K_m$  form is associated with putative plasma membrane  $(F_1DF_3)$  (Table I). Treatment of  $F_2$  with digitonin also yields a fraction  $(F_2DF_3)$  with an increased density, whose nucleotidase has a  $K_m$  essentially unchanged from the parent  $F_2$  fraction (Table I).

# Effects of membrane solubilization on the enzyme forms

There are two likely explanations for the forms observed in the different

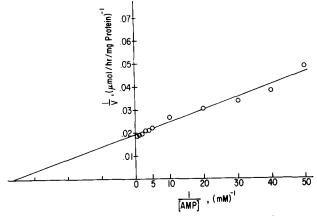


Fig. 2. Lineweaver-Burk plot of F<sub>2</sub> membrane fraction 5'-nucleotidase kinetics.

TABLE I  $K_{\mathbf{m}}$  VALUES FOR 5'-NUCLEOTIDASE OF MEMBRANE AND SOLUBILIZED FRACTIONS OF NORMAL RAT MAMMARY GLAND

Membrane fraction	$K_{\mathbf{m}}$ ( $\mu$ M)		
F <sub>1</sub> high K <sub>m</sub>	130		
F <sub>1</sub> low K <sub>m</sub>	22		
F <sub>1</sub> + deoxycholate	28		
F <sub>1</sub> + Triton X-100	23		
F <sub>1</sub> + digitonin:			
High K <sub>m</sub>	120		
Low Km	18		
F <sub>1</sub> DF <sub>1</sub>	28		
F <sub>1</sub> DF <sub>1</sub> + Triton X-100	21		
F <sub>1</sub> DF <sub>3</sub>	120		
F <sub>1</sub> DF <sub>3</sub> + deoxycholate	37		
F <sub>1</sub> DF <sub>3</sub> + Triton X-100	34		
F <sub>2</sub>	27		
F <sub>2</sub> DF <sub>3</sub>	26		

membrane fractions. (1) There are two different types of 5'-nucleotidase molecules with different properties. (2) There is one type of nucleotidase molecule, but its properties vary with its mode of association with the membrane and with the properties of the membrane. If the latter is true, it should be possible to eliminate the differences between the forms by disrupting the membrane structure, e.g. by solubilizing the enzyme. Fig. 3 shows the 5'-nucleotidase activity of the  $F_1$  fraction solubilized by deoxycholate. More than 80% of the enzyme was soluble, and the curves for solubilization of the two enzyme forms, as assayed at 0.1 mM (Fig. 3) and 1.0 mM (not shown) substrate, were essentially identical. Kinetic studies of the soluble enzyme indicated that the low  $K_m$  enzyme was by far the predominant species (Table I).

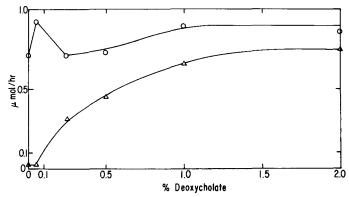


Fig. 3. Solubilization of 5'-nucleotidase activity with deoxycholate.  $F_1$  membranes were suspended in 50 mM Tris (pH 8.0) containing various concentrations of deoxycholate. After a 1 h incubation at  $37^{\circ}$ C, the suspensions were centrifuged at  $40\,000 \times g$  for 2.5 h. The supernate was assayed for 5'-nucleotidase activity in the presence of 0.1 mM AMP (containing 1-5  $\mu$ Ci/ml [ $^{3}$ P]AMP). This is referred to as the solubilized activity ( $^{\triangle}$ —— $^{\triangle}$ ). The total activity ( $^{\bigcirc}$ —— $^{\bigcirc}$ ) represents the activity present before the centrifugation.

A dramatic reduction in  $K_{\rm m}$  was observed when purified  $F_1DF_3$  was treated with deoxycholate (Table I). This behavior is consistent with the premise that the properties of the enzyme in the  $F_1DF_3$  fraction are altered by its association with the membrane, since these properties more closely resemble the low  $K_{\rm m}$  form when the membrane is solubilized.

Similar results were obtained when the non-ionic detergent Triton X-100 was used. The  $K_{\rm m}$  values are not significantly different from those found with deoxycholate (Table I). When membranes containing only the low  $K_{\rm m}$  form of nucleotidase (e.g.  $F_1DF_1$ ) are solubilized in either deoxycholate or Triton X-100, the  $K_{\rm m}$  values for the enzyme are not substantially changed (Table I).

# Temperature dependence of the nucleotidase activity

As a further indication of the influence of the membrane-enzyme association the temperature dependence of the nucleotidase in the various membrane fractions was examined. Arrhenius plots for the  $F_1$  fraction assayed with 1.0 mM AMP showed two breaks (Fig. 4). The  $F_1$  fraction assayed at 0.1 mM AMP and

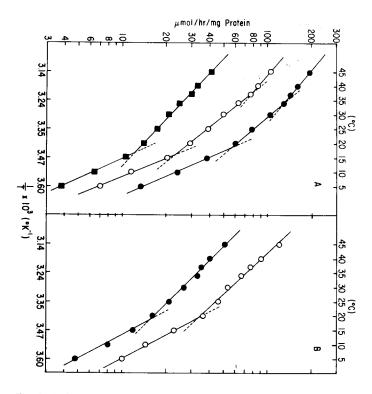


Fig. 4. Arrhenius plots for mammary membranes and Triton-solubilized fractions. (A) Membranes were incubated at the appropriate temperature in assay media for 10 min prior to assaying at the same temperature.  $F_1DF_3$  membranes ( $\bullet$ ) and  $F_1$  membranes ( $\circ$ ) were assayed with 1.0 mM ATP, while  $F_2$  membranes ( $\bullet$ ) were assayed with 0.1 mM AMP. (B) Membranes were suspended in 10 mM Tris (pH 8.0) containing 1% Triton X-100 and incubated for 1 h at 37° C. The solubilized membranes were then incubated at the appropriate temperature in assay media for 10 min prior to assaying at the same temperature.  $F_1$  Triton-solubilized membranes ( $\circ$ ) were assayed with 1.0 mM AMP, and  $F_2$  Triton-solubilized membranes ( $\circ$ ) were assayed with 0.1 mM AMP.

TABLE II

TRANSITION TEMPERATURES AND ACTIVATION ENERGIES FOR NUCLEOTIDASE IN VARIOUS MEMBRANE AND SOLUBILIZED FRACTIONS

All values reported are averages of two or three separate determinations on different membrane preparations with standard deviations of less than 10% of the reported values. The blanks indicate no treatment or a failure to observe a corresponding value for that sample.

Fraction	Treatment	AMP (mM)	Activation energy (kcal/mol)			Transition temperature	
			$E_{a1}$	$E_{\mathbf{a}2}$	E <sub>a3</sub>	$\overline{T_1}$	$T_2$
F <sub>1</sub>	_	1.0	17.8	10:5	7.4	15.3	36.7
F <sub>1</sub>	_	0.1	13.4	7.4	_	18.0	_
F <sub>2</sub>	_	0.1	14.3	8.0		18.2	_
F <sub>1</sub>	digitonin	0.1	14.8	7.3	_	18.3	
F <sub>1</sub>	digitonin	1.0	17.7	9.2	6.8	17.6	37,7
$F_1DF_3$	_	1.0	16.9	10.1	7.3	18.2	34,4
$\mathbf{F_1}$	deoxycholate	0.1	14.0	7.7	_	19.2	
F <sub>1</sub>	deoxycholate	1.0	14.4	9.0	_	19.2	<del></del>
$\mathbf{F_2}$	deoxycholate	0.1	14.0	8.6	_	19.1	
$\mathbf{F}_{1}^{\mathbf{z}}$	Triton X-100	1.0	13.6	7.6		20.7	_
F <sub>1</sub>	Triton X-100	0.1	13.2	7.1	_	19.2	_
F <sub>2</sub>	Triton X-100	0.1	13.7	8.0	_	18.8	_

the  $F_2$  fraction gave only one break (Table II). Treatment of  $F_1$  with the digitonin did not alter this behavior, except for a slight shift in the lower transition temperature. Purified  $F_1DF_3$  also showed both transitions (Fig. 4). When  $F_1$  is treated with deoxycholate or Triton X-100, the high temperature transition observed in the presence of 1.0 mM AMP is lost (Fig. 4 and Table II).

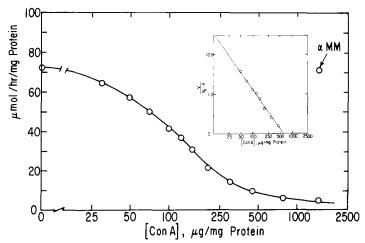


Fig. 5. Concentration dependence of concanavalin A inhibition of  $F_1$  membrane 5-nucleotidase assayed with 1 mM AMP. Membranes, prepared as described in Experimental procedures and Fig. 1, were incubated 10 min with concanavalin A before assaying with 1 mM AMP (containing 1-5  $\mu$ Ci/ml [ $^{32}$ P]-AMP). 50 mM  $\alpha$ -methylmannoside ( $\alpha$ -MM) was added to a tube containing the highest concanavalin A concentration to show reversibility of the reaction. The  $\alpha$ -methylmannoside was added after incubation with concanavalin A and the membranes incubated an addition 10 min before assaying. The inset shows a Hill plot for concanavalin A inhibition of  $F_1$  membrane 5'-nucleotidase (n, 1.4).

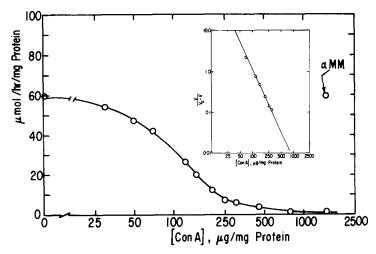


Fig. 6. Concentration dependence of concanavalin A inhibition of  $F_1$  membrane 5'-nucleotidase assayed with 0.1 mM AMP. Membranes were prepared, incubated and assayed as in Fig. 3 except that 0.1 mM AMP (containing 1—5  $\mu$ Ci/ml [ $^{32}$ P]AMP) was used. The inset shows a Hill plot for concanavalin A inhibition of  $F_1$  membrane 5'-nucleotidase in the presence of 0.1 mM AMP (n, 2.2).

There is essentially no change in the low temperature transition resulting from detergent treatments. Thus the loss of the high temperature transition on detergent treatment is correlated with the change in  $K_{\rm m}$  previously noted. There is no obvious relationship between the low temperature transition and  $K_{\rm m}$ . Data for the transition temperatures and activation energies of the various membrane and solubilized species (Table II) show the similarity of the solubilized forms to the form observed in membranes exhibiting a low  $K_{\rm m}$  enzyme.

### Cooperativity of the concanavalin A inhibition

Treatment of  $F_1$  fractions with concanavalin A caused essentially complete inhibition of either the high  $K_m$  (Fig. 5) or low  $K_m$  (Fig. 6) enzyme. Both were readily reversible by  $\alpha$ -methylmannoside, indicating a specific interaction of the concanavalin A with carbohydrate. However, the degree of cooperativity was different for the two. A Hill coefficient of 1.4 was seen with the high  $K_m$ 

TABLE III

COOPERATIVITY IN THE CONCANAVALIN A INHIBITION OF NORMAL RAT MAMMARY MEMBRANE 5'-NUCLEOTIDASE

Membrane fraction	Treatment	AMP (mM)	Hill coefficient	
F <sub>1</sub>	<del>-</del>	0.1	2.3	
F <sub>1</sub>	_	1.0	1.4	
F <sub>2</sub>	_	0.1	1.9	
F <sub>1</sub> DF <sub>3</sub>	_	1.0	1.4	
F <sub>1</sub> DF <sub>3</sub>	deoxycholate	0.1	2.1	
F <sub>1</sub> DF <sub>3</sub>	deoxycholate	1.0	2.1	
F <sub>1</sub>	deoxycholate	1.0	2.2	
F <sub>1</sub>	Triton X-100	1.0	2.1	

enzyme form, whereas a Hill coefficient of 2.2 was observed for the low  $K_{\rm m}$  form. Purified  $F_1DF_3$ , which shows a high  $K_{\rm m}$  enzyme, also exhibits decreased cooperativity (Table III).

When membranes containing the high  $K_{\rm m}$  enzyme were solubilized in deoxycholate, there was a pronounced increase in cooperativity (Table III) along with the decrease in  $K_{\rm m}$  previously described. In addition the  $F_2$  membrane fraction, which has only the low  $K_{\rm m}$  enzyme, exhibits cooperativity for the concanavalin A inhibition (Table III).

### Discussion

Rat mammary gland microsomes can be separated by flotation into two fractions ( $F_1$  and  $F_2$ ) which contain plasma membrane material [13]. Both are enriched in 5'-nucleotidase, (Na+ K+)-ATPase, cholesterol, sialic acid and galactosyltransferase. The last observation suggests the presence of Golgi in the preparations, since galactosyltransferase is enriched in the Golgi apparatus. If F<sub>1</sub> is treated with digitonin and refloated on a gradient with the same composition, a portion of the membranes moves to a higher density. This material (F<sub>1</sub>DF<sub>3</sub>) is highly enriched in nucleotidase, sialic acid and cholesterol but has no galactosyltransferase [14]. The portion of the digitonin-treated fraction which retains the original density upon recentrifugation, called F<sub>1</sub>DF<sub>1</sub>, is enriched in galactosyltransferase but depleted in nucleotidase, cholesterol and sialic acid. The F<sub>1</sub>DF<sub>3</sub> and F<sub>1</sub>DF<sub>1</sub> fraction are therefore considered to be purified plasma membranes and Golgi fragments, respectively. If F<sub>2</sub> is treated with digitonin, a portion of the sample migrates to a higher density. This F<sub>2</sub>DF<sub>3</sub> is enriched in nucleotidase, sialic acid, cholesterol and galactosyltransferase, suggesting that it is a plasma membrane containing galactosyltransferase (unpublished data). This proposal is supported by the similarity of sialoglyprotein profiles for  $F_2DF_3$  and the milk fat globule membrane of rat milk (unpublished data).

Kinetics of the nucleotidase in the various membrane fractions indicate the presence of two enzyme forms in  $F_1$ . These forms segregate with the  $F_1DF_1$  and  $F_1DF_3$  fractions. Thus a question arises as to the nature of the differences between the two forms. One obvious possibility is that they represent two different enzyme molecules (isoenzymes) which are separately synthesized and incorporated into the different types of membranes. We feel this is unlikely because the high  $K_m$  form of the enzyme can be converted to a form with properties closely resembling the low  $K_m$  form by treatment with detergents. The conversion can be accomplished by two different detergents with different solubilizing properties and different effects on membrane enzymes. In addition three different properties of the enzyme are coordinately altered.

In view of these results a more likely explanation is that there is one type of enzyme whose properties are dependent on its association with the membrane. According to this proposal the increased  $K_{\rm m}$  and decreased cooperativity would be a result of restrictions placed on the enzyme by its association with other membrane components. This behavior would correspond with the observation of a new transition in the Arrhenius plots, indicative of the presence of another form of the enzyme. This transition is eliminated by detergent treatments, sug-

gesting that it is dependent upon interactions that are disrupted by the detergents. The transition near  $20^{\circ}$ C is not influenced by detergents and shows no correspondence to changes in the other enzyme properties. These results suggest that this transition is probably a function of a change in the enzyme molecule itself, although other tightly bound species could be involved. The fact that the transition near  $20^{\circ}$ C is not substantially altered by detergent treatments is another indication that the detergents are not substantially altering the structure of the enzyme molecule, even though they are changing the  $K_{\rm m}$  and the response to concanavalin A.

Thus we feel there is no necessity for postulating the occurrence of isoenzymes to explain the differences between enzymes of  $F_1DF_3$  and  $F_1DF_1$ . However, this should not be taken to indicate that there are necessarily no molecular differences between the two enzyme forms. Since the nucleotidase is a glycoprotein, it is quite possible that the two forms of the enzyme, which are found in different types of membranes, could have variations in carbohydrate content which would give them different physical properties without significantly affecting their  $K_m$  values or the cooperativity parameter.

The molecular basis for the difference in the properties of the two forms is still unclear. It is interesting to note that the two purified membrane fractions (F<sub>1</sub>DF<sub>3</sub> and F<sub>2</sub>DF<sub>3</sub>) which have characteristics of plasma membranes (high nucleotidase, cholesterol and sialic acid) have different forms of the enzyme. The enzyme and density differences of these fractions indicate that they must have differences in protein or lipid compositions. Preliminary experiments using fluorescent probe techniques do not show prominent differences in microviscosities which can be related to the enzyme behavior (Doss, R.C. and Smith, W.B., unpublished data). This is supported by observations on membrane cholesterol, which is a primary modulator of membrane fluidity [17]. Cholesterol values are high for F<sub>1</sub>DF<sub>3</sub> and F<sub>2</sub>DF<sub>3</sub>, which have different nucleotidase forms. In addition F<sub>1</sub>DF<sub>1</sub>, which has low cholesterol, and F<sub>2</sub>DF<sub>3</sub>, which has high cholesterol, apparently have the same enzyme form. Thus there appears to be no relationship between cholesterol and the kinetics of the membrane enzyme. These results suggest that the observed behavior of the enzyme in these membranes is dependent on specific enzyme-lipid or enzyme-protein interactions rather than on the bulk membrane lipid phase. Nucleotidase isolated from rat liver shows a specific association with sphingomyelin [18]. Whether such an association exists in the mammary membranes requires further investigation.

In previous studies we have shown that the cooperativity of the nucleotidase inhibition by concanavalin A is dependent on the 'state' of the membrane [7]. We postulated that the cooperativity was dependent on the multivalency of the concanvalin A and mobility of the enzyme in the membrane [16]. The present results are consistent with this hypothesis, if it is also assumed that decreased  $K_{\rm m}$  and increased cooperativity correlate with a release of restraints in the membrane imposed by specific lipid-enzyme or protein-enzyme interactions, as suggested earlier. Further investigations by membrane modification and reconstitution techniques should aid in defining these parameters and increase our understanding of the behavior of the nucleotidase and membrane enzymes in general.

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